

in homes

#### Science content

Radioactivity, radioactive decay, background radiation, alpha radiation.

#### Science curriculum links

- AT 1 Exploration of science AT 8 Explaining how materials
- behave

#### Syllabus links

- GCSE Science, Biology, Physics,
- Geography
- Sixth-form General Studies
- Cross-curricular themes
- Health Education

Lesson time

1 to 2 hours

(homework possible)

Links with other SATIS materials

- 1005b Radon an investigation 204 Using Radioactivity
- 807 Radiation how much do you
- get?

SATIS Audiovisual Radiation around us

NERIS

Search on

RADON or on RADIATION and ENVIRONMENTAL HEALTH

Author Anabel Curry

First published 1991

# SUMMARY

A radioactive gas called radon finds its way into homes. The unit tells how scientists in the UK have become more aware of the problem and the risks it poses to health.

# **STUDENT ACTIVITIES**

- □ Reading, answering questions based on the text, interpretation of graphical data.
- $\Box$  Discussion of the issues involved, reporting back to the class.
- □ Follow-up work which could involve creating community awareness of the problem designing a poster, writing letters, giving talks etc. or measuring radon activity as described in SATIS unit 1105b.

# AIMS

- $\Box$  To link with work on ionising radiations
- □ To show how scientific understanding of the radon problem has developed
- □ To increase awareness of the danger to health of radon and cigarette smoking
- □ To consider and discuss the social, moral, legal and economic implications of the radon problem
- □ To provide an opportunity for students to interpret their scientific knowledge to the wider community
- □ To provide students with the opportunity to take action on an issue concerning public health

# USE

- □ The unit may be used as part of work on ionising radiations and their effects on living organisms. There are also links with work on health and pollution.
- □ The reading and text-related questions may be set for homework. The discussion and follow-up work may be omitted.
- $\Box$  Follow-up work suggests involving students with the wider community.

# **Teaching notes**

The focus of attention on naturally-occurring radiation is turning from gamma radiation (which tends to pass through the body) towards alpha which, when lodged inside the body, is much more damaging. Radon is more widespread and poses a greater hazard than was once supposed.

For simplicity, the unit focuses on 'radon', actually radon-222, which occurs in the uranium-238 decay series. Another isotope of radon, radon-220, is known as thoron and occurs in the thorium decay series.

The part of the uranium decay series containing radon is:

 $^{234}U_{\sim} \rightarrow ^{230}Th_{\sim} \rightarrow ^{226}Ra_{\sim} \rightarrow ^{222}Rn_{\sim} \rightarrow ^{218}Po_{\sim} \rightarrow ^{214}Pb$ 

A becquerel is a disintegration per second. Thus a radon activity of 1 Bq/m<sup>3</sup> means that on average 1 atom of radon in a cubic metre of air decays every second. The relationship between 'activity' and 'dose equivalent' is complex. Dose is measured in sieverts. Further details are given in SATIS 807, *Radiation – how much do you get?* 

# Follow-up work

This topic gives students a genuine opportunity to disseminate information about the hazards posed by radon in homes. Students may come up with suggestions for designing posters, writing for the school magazine, giving talks to the school or possibly making an impact on the community at large.

They may go on to investigate radon levels around them as described in SATIS unit 1105b, *Radon – an investigation*.

# **Other resources**

The National Radiological Protection Board (NRPB) is an independent statutory body. It publishes booklets and posters, some specifically designed for educational use. A new leaflet, *Radon-Questions and Answers*, provides simple and clear information about radon. Details are available from the Information Officer, NRPB, Chilton, Oxfordshire OX11 0RQ.

A booklet *The Householders' Guide to Radon* Department of Environment, HMSO, 1989, gives advice on radon-proofing homes.

# Answers to the questions

In their understanding of radioactivity, students need to distinguish between 'radiation' (such as alpha, beta and gamma) and radioactive material (which gives off radiation).

The questions are based on information given in the text.

- Q1 Radon is a gas with no smell, taste or colour.
- Q2 Stanley had become 'radioactive', implying contamination with radon (and its decay products). The radiation given off by the contamination set off the alarms at his place of work. The search for the source led to his home.
- Q3 At 20 Bq/m<sup>3</sup>, the risk of developing cancer is very low. (Casual passive smoking may pose a greater risk.)
- Q4 (a) About 135 (b) 2 packs
- Q5 Radon decays emitting an alpha particle which may damage living cells and lead to cancer.
- Q6 (a) July (b) living rooms (c) less
- Q7 Pressure inside homes is less than outside (because heating makes warm air rise and escape) and radon moves in. Draught stripping in modern homes prevents radon escaping.
- **Q8** Explain to him/her about the health hazard and that not all homes are at risk. Suggest that the radon level is measured. (Advice is available from the local authority or NRPB.) If the level is above 200 Bq/m<sup>3</sup>, modifications should be carried out to the building to allow radon trapped under the house to escape. Seal around cracks and pipes; lay a plastic barrier over the ground floor; provide regular ventilation of the house.

#### Acknowledgements

Jon Miles of NRPB and H. Eijkelhof of the University of Utrecht read and commented on the trial version.

Figure 2 is adapted from data in *Consumer Reports* July 1987.

Figure 3 after Green, Lomas and O'Riorden, NRPB.

Headlines from *The Independant*, 26 October 1989 and 20 January 1990.



# Alarm bells ring

Stanley didn't know it when he dressed for work on 2 December 1984 but this was the day his name would go around the world. He was to become famous. Stanley Watras was an ordinary sort of guy, a worker at a nuclear power plant near Philadelphia in the USA. In 1984, he became the centre of a mystery, a problem that nobody had suspected existed.

On that December day Stanley walked into work and set the alarm bells ringing. Officials were perplexed. Was 'radiation' escaping from the plant or was there something wrong with the alarm system?

Stanley set off the alarms every time he went into work. Nobody else did.

Stanley was radioactive and it wasn't anything to do with the nuclear power plant. But had he *not* worked in the nuclear power industry, his problem might never have been known.

It took two weeks to track down the reason. Stanley's home had been built over a vein of rock containing uranium.

Uranium is a radioactive element. During its decay, it forms a radioactive gas called **radon**. Radon has no smell, taste or colour. It had seeped through the rocks into Stanley's home. The amount of radon had gradually built up and Stanley had been breathing it in.

The mystery was solved!

#### Radon: the search starts

Was Stanley Watras the only person with a high level of radon? Nobody knew.

Scientists got out their geological maps and looked to see who else might be living on top of uranium-rich rocks. They tested homes for radon and found it. Warnings went out to Americans believed to be at risk. They were advised to seal cracks in their basements and ventilate their homes.

As more homes were tested, radon was found far from any uranium-rich rocks. In fact, uranium occurs all over the Earth's crust and radon is almost everywhere; but its concentration is unpredictable. If radon reaches the surface soil it quickly disperses into the air. But being a gas, it can seep through cracks in the rocks and build up in individual homes. This unit contains material for reading, answering questions and opportunities for group discussion.

It may be used in association with SATIS unit 1105b, *Radon* - an investigation, which describes how to measure the levels of radon gas in homes.

**Q1** Why did Stanley Watras not suspect his home had radon gas in it?

**Q2** What led to the discovery of radon in Stanley's home?

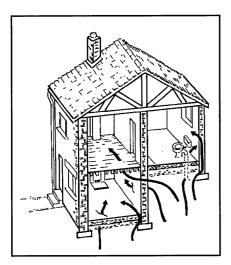


Figure 1 How radon seeps into a house

In the UK, scientists already knew that radon was a risk for miners in Cumbria and Cornwall. They had started looking for radon in homes in areas of granite rock, like Cornwall, Devon and around Aberdeen in 1980. They found it – sometimes in dangerous concentrations. But as in the US, amounts varied from home to home, depending on where and how the homes were built and ventilated.

The level of radioactivity of radon in the air is measured in **becquerels per cubic metre** ( $Bq/m^3$ ). The average level of radon in UK homes is 20 Bq/m<sup>3</sup>. In the US it is about 50 Bq/m<sup>3</sup>. Outdoors the level of radon in air is a few becquerels per cubic metre. Radon is found in rocks, soil and dissolved in drinking water too.

Scientists now believe that radon is a far greater risk to health than any other environmental pollutant – such as smoke and carbon monoxide in the air, pesticides and nitrates in food and water. Radon causes lung cancer. Only cigarette smoking is more dangerous!

The graph shows how your risk of lung cancer from radon compares with the risk you take with smoking. Stanley Watras was said to be receiving a dose of radiation from radon equivalent to smoking 135 packs of cigarettes a day!

lifetime exposure to radon (Bq/m<sup>3</sup>) 40 80 average home in UK = 20 Bq/m<sup>3</sup> 160 1 þack a day 400 800 2 packs a day 1600 21/2 packs a day 4000 3 packs a day 8000 1000 +4 packs a day 800 200 0 400 600

Figure 2 Radon and cigarettes: the death rate from lung cancer

estimated death rate per 1000 people exposed

**Q3** The average home in the UK has a radon level of 20  $Bq/m^3$ . Does the graph suggest that this level is likely to lead to lung cancer?

Q4 A few homes have levels of radon as high as 800 Bq/m<sup>3</sup>. Out of 1000 people living in such homes, (a) how many may die of lung cancer caused by radon? (b) how many packs of cigarettes a day give the same death rate from lung cancer? Like any radioactive element, radon decays. Each atom emits an alpha particle. Radon decays to give other radioactive elements called **radon daughters**. They also emit alpha particles. When you breathe in radon and its daughters, some of the daughters stick inside the lung. If an alpha particle strikes a living cell, it may damage it. Damaged cells can lead to cancer.

Lung cancer from smoking kills about 40 000 people a year in Britain. It is estimated that radon causes a further 2500 lung cancer deaths. For smokers radon increases an already large risk.

Radon gas may be the cause of some cases of other forms of cancer, such as myeloid leukaemia. Scientists are carrying out further studies to see if this is so.

Radon enters houses because air pressure outside is a little higher than that inside. This is due warm air in the building rising. Wind flowing across chimneys and windows causes suction. Radon seeps in through cracks in walls and floors and often around the holes for water and waste pipes. Radon levels tend to be high in homes with basements.

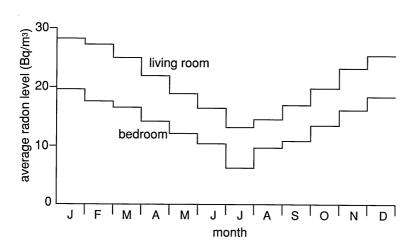


Figure 3 Variation in radon levels in living rooms and bedrooms during the year

- *Q5* Draw a flow chart or cartoon explaining how radon causes lung cancer.
- **Q6** Study the graph showing radon levels in a typical British house.

(a) In which month is the radon level lowest? Suggest why this is.

(b) Where are people exposed to higher levels of radon, in the living room or bedroom?

(c) Would a family living on the fourth floor of a block of flats be exposed to more or less radon than a family living in a typical house?

#### Facts about radon

 Radon is a naturallyoccurring radioactive gas that is inactive chemically but soluble in water.

 Radon rises from the ground and collects in buildings.
 Householders who have levels of radon activity greater than 200 Bq/m<sup>3</sup> should 'radon proof' their homes.

 The radioisotope known as 'radon' is radon-222 and is part of the uranium decay chain.

• When radon-222 decays it emits an alpha particle.

• The half-life of radon-222 is 3.8 days.

 Radon is a gas, but radon daughters are solid isotopes, polonium-218, polonium-214, lead-214 and bismuth-214.

 In fact radon gas accounts for only 1% of the radiation dose from radon. Most comes from radon daughters. They account for more than half the dose of natural radiation that people receive.

Type of rock	Average radon concentration in homes in Bq/m <sup>3</sup>
Granites	
SW England	285
Scotland	21
Sandstone	50
Chalk	26
Clays	15

# Reducing the risks from radon

The Swedish Government introduced limits to radon in homes as long ago as 1980. The US recognised it had a similar problem after Stanley set off the alarms.

In 1987 the British Government provided help for householders if radon levels in their homes were above 400 Bq/m<sup>3</sup>. With greater awareness of the dangers, the 'action level' is now set at 200 Bq/m<sup>3</sup>.

Only a few years ago, the radon problem was thought to be limited to areas of granite rock in Devon, Cornwall and Scotland. The results of surveys show that high levels of radon may occur elsewhere. The search for radon in homes is widening to Somerset, Northamptonshire, Derbyshire, the Highlands and Islands of Scotland and Grampian region.

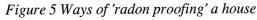
In an area of high radon levels not all homes will be affected. The level of radon depends on the materials from which the home is built and the way it is heated and ventilated. And levels can vary widely from day to day in just one house as the graph shows. So the level of radon inside a house needs to be taken over a period of time.

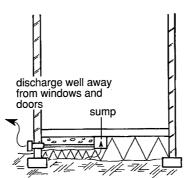
Householders at risk are offered a free survey. Those with high levels of radon in their homes are notified. They may apply for grants to radon-proof their homes. At present only people on low incomes receive money for improvements.

Radon-proofing usually involves putting in a ventilation system with a fan under the floor to allow radon to escape. Cracks and gaps around pipes must be sealed. A plastic sheet may be laid over wooden floors to stop radon from rising into the house. Modifications can cost  $\pounds$ 500 to  $\pounds$ 1,000.

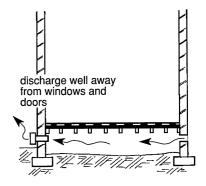
**Q7** Explain why radon levels can build up in homes.

**Q8** Make a list of practical advice you would give to an aunt, uncle or friend who lives in an area where radon levels in homes may be high.





A house with a solid floor and a radon 'sump' dug underneath. A fan draws the radon away.



A house with a wooden ground floor and a fan to ventilate the area underneath.

# radon level

Figure 4 Graph showing how radon levels in a home may vary from week to week

# The problem no one wants to face

In both the USA and the UK many of the people with high radon levels in their homes have done nothing about it.

In Washington DC in the USA, a local television station promoted the sale of radon detecting kits through supermarkets. 60 000 were sold. The results showed that 14 000 homes had high radon levels. After 18 months less that 2500 homes had done anything to reduce the risk.

The scientific detective story that started with Stanley Watras is still unfolding. How long will it take before people in the UK recognise the dangers of radon? Students in schools and colleges have an important part to play in helping others to understand the problem.

# Radon gas is twice as as thou Cancer-causing sas 'widespread'

Headlines from The Independent, 26 October 1989 and 20 January 1990

# Questions for group discussion

- a If homes have radon levels above 200 Bq/m³, householders are advised to radon proof their homes. How many people per 1000 may die at this level? (Look back at the graph on page 2.) How does it compare with the risk smokers take? Does this level of risk seem reasonable?
- **b** When advised they have high levels of radon many householders do nothing to radon-proof their homes. Why do you think this is?
- *c* Who should pay for surveys and building work to 'radonproof' homes – homeowners, insurance companies, local or national government? If not the homeowners, would other people mind paying extra premiums or taxes to cover the bill?
- *d* What sort of laws could the Government pass to limit the danger of radon in homes?
- *e* How should the public be made more aware of the dangers of radon? Include in your suggestions something you can carry out for yourself.

# Follow-up

Carry out one or more of the suggestions you made in *e* above.

#### Organising the discussion

- Work in groups of 3 to 6 students.
- Appoint someone to chair the group and to report back to the class if required.
- Note down the answers you decide on. You may need to use your ideas later.

Answers to questions are given in the *Teachers' Notes*.

SATIS unit 1105b, *Radon – an investigation,* describes how to measure the levels of radon around you.



#### Science content

Radioactivity, radioactive decay, background radiation, alpha radiation.

#### Science curriculum links

- AT 1 Exploration of science AT 8 Explaining how materials
- behave

Syllabus links • GCSE Science, Physics,

#### Lesson time

3 hours

- (2 lessons and 2
- homeworks)

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ENVIRONMENTAL HEALTH

# SUMMARY

The unit provides information for students to design and carry out an investigation to monitor radon levels in buildings or in the soil. The experiment requires students to count the alpha tracks with a microscope. There is a free service to schools for processing the detector, TASTRAK. Students may contribute their results to a national survey.

# **STUDENT ACTIVITIES**

- □ **Part A** Planning the investigation. (Suitable for homework)
- □ **Part B** Making a radon detector, setting up the experiments, sending the TASTRAK for processing.

Analysing TASTRAK after processing; recording data; writing a report.

□ **Part C** Data interpretation. Questions.

# AIMS

- $\Box$  To link with work on ionising radiation
- □ To develop awareness of radon as a radioactive gas, an alpha emitter and a major contributor to our background exposure to radiation
- □ To provide an opportunity for students to plan and carry out a scientific investigation
- $\Box$  To contribute to a national survey of radon levels

# REQUIREMENTS

- □ clean yoghurt pots, cylindrical, (50 mm diameter and 70 mm high) from brands such as Chambourcy, Eden Vale, Ski, Sainsburys, etc.
- □ Blu-Tack, clingfilm, elastic bands
- □ pieces of TASTRAK (from address given here)
- $\square$  microscopes (magnification  $\times$  200)
- □ scale for calculating field of view [Philip Harris Educational B 32200/9 (pack of 10 at £7.91)]
- □ small plastic bags and cooking foil (for keeping the TASTRAK while setting up the experiment)

The booklet, TASTRAK and the FREE PROCESSING SERVICE is available from Track Analysis Systems Limited, H H Wills Physics Laboratory, University of Bristol, Tyndall Avenue, Bristol, BS8 1TL. Tel. 0272-260353

Authors Geoffrey Camplin Denis Henshaw

First published 1991

# Teaching notes

The unit contains simple experiments to measure the activity of radon. They have been thoroughly tested with pupils aged between 7 and 16 years. It is suggested that students should have prior understanding of ionising radiation and the radon problem, possibly through using SATIS 1105a, *Radon in Homes*.

An ordinary yoghurt pot must be used with TASTRAK if an absolute radon activity is to be determined, since this type of pot has been calibrated by the National Radiological Protection Board. The calibration factor is given in this unit.

Pages 1 to 3 could be read in advance (perhaps for homework). Students may wish to negotiate with other groups in order to plan an effective survey. This could be a teacher-led class activity.

If students have difficulty in calculating the field of view of the microscope, the teacher may wish to give students this figure.

You are advised to have the TASTRAK processed professionally with the FREE SCHOOLS' PROCESSING SERVICE at Bristol University by Track Analysis Systems Limited (TASL). Send a stamped addressed envelope. This is financed by UKAEA under the guidance of TASL at Bristol University's Physics Department.

Please use this service.

The Exploratory in Bristol and Portway School are compiling a national map of radon levels. Data from your students' work will be welcomed. Please fill in the results together with a six figure Grid Reference of where they were taken on the form provided and post it to Dr G. C. Camplin, Portway School, Penpool Lane, Shirehampton, Bristol BS11 0EB. (Help for working out Grid References is given on Ordnance Survey maps or in geography textbooks.)

The detailed analysis of alpha particle tracks in TASTRAK can be found in the references. Only a basic description is given in the unit because of the highly mathematical nature of the analysis. These experiments have been taken from the booklet *TASTRAK: A New Plastic Track Detector for Teaching Radioactivity in Schools* – see Further Resources.

#### Further resources and references.

- Camplin G. C., Henshaw D. L., Lock S. and Simmons Z., 1988 *Phys. Educ.* 23 p212–217 'A National Survey of Alpha Particle Radiation' (London: Institute of Physics).
- 2 Fews A. P. and Henshaw D. L., 1982 Nucl. Instrum. Methods 197 p517–529 (North Holland) 'High Resolution Alpha Particle Spectroscopy using CR-39 Plastic Track Detector'.
- 3 Henshaw D. L. and Camplin G. C., *TASTRAK: A New Plastic Track Detector for Teaching Radioactivity in Schools*, London: Institute of Physics (1987).

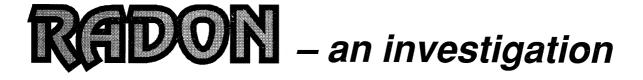
#### Answers to the questions

- Q1 Radon disperses into the air at the surface.
- Q2 The histogram shows a peak frequency between 100 and 130 Bq/m<sup>3</sup>. The frequency drops off rapidly thereafter. A few homes had radon activities above the 'action level' of 200 Bq m<sup>3</sup>. No school had a zero result.

Students should add a comment about their own result – whether it seems low, average or high.

- Q3 Portway School shows a peak frequency of around 50 Bq/m<sup>3</sup> with no homes above the action level. Brookside Primary School survey included more homes and follows the same pattern. However several (29) homes have a radon activity above the action level, the highest being around 1079 Bq/m<sup>3</sup>.
- Q4 Definitions may be found in a good physics textbook.
- Q5 Alpha particles can travel no more than 50 mm in air. They will be unable to reach the TASTRAK through the mouth of the yoghurt pot 70 mm high (even if some pass through the clingfilm). They are unable to pass through the sides of the pot. Radon can diffuse in through clingfilm. The TASTRAK therefore detects only the alpha particles that result from the decay of radon (and radon daughters).

Price for a pack of 30 pieces of TASTRAK is about £20.00 including VAT. Please ring for up-to-date prices. (TASTRAK scribed with one centimetre squares which can be used in a projector for counting is available at additional cost.)



This unit describes how you may investigate how much radon is around you. And you can add your results to a national survey. *Read through parts A and B before you plan your work.* 

# Part A – What do you need to know?

# Radon

There are many natural sources of radiation. But it is the **radioactive** gas, **radon**, which is particularly dangerous to health. It rises from the ground and collects in buildings. You can detect it by the **alpha radiation** it emits when it decays.

The experiments in this unit use a plastic called TASTRAK as detector. When an alpha particle enters the surface, it creates a trail of damage along its path. You cannot see the tracks until the plastic has been processed in an etching solution.

The tracks are up to 30 micrometres long and you need a microscope to see them. Alpha particles enter the plastic from different directions giving the tracks different shapes and sizes.

Counting the tracks tells you how many alpha particles hit the plastic and gives you a measure of the amount of radon around.

TASTRAK must be kept in the metallised plastic bag in a freezer or refrigerator until it is ready for use. This bag is 'radon proof' – alpha radiation from the environment cannot get in through the bag.

# Planning your investigation

Here are some points to consider.

- $\Box$  How many detectors can your group or class make?
- $\Box$  Do you wish to measure radon in soil or in buildings or both?
- □ Do you wish to link your measurement to those of other groups in your class? Would it be better to survey just one spot or survey a building or locality?

Read through the instructions in part B before you draw up your plans.

**Part A** explains about radon and how TASTRAK may be used to detect the alpha radiation it emits.

Part B tells you how to make and use a radon detector.

**Part C** looks at data obtained by students in other schools.

Page 6 contains more information and questions.

SATIS 1105a, *Radon in Homes*, provides introductory information to the problem of radon.

#### Facts about radon

 Radon is a naturallyoccurring radioactive gas that is inactive chemically but soluble in water.

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 The radioisotope known as 'radon' is radon-222 and is part of the uranium decay chain.

• When radon-222 decays it emits an alpha particle.

• The half-life of radon is 3.8 days.

 Radon is a gas, but radon daughters are solid isotopes, polonium-218, polonium-214, lead-214 and bismuth-214.

 Radon gas accounts for only 1% of the radiation dose from radon. Most comes from radon daughters. They account for more than half the dose of natural radiation that people receive.



- o a clean yoghurt pot
- Blu-Tack
  clingfilm
- piece of TASTRAK
- o elastic band
- o compass point or drawing pin
- o plastic bag and cooking foil

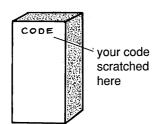
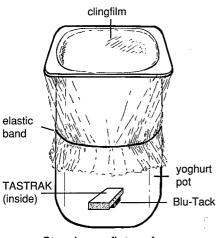


Figure 1 Preparing the TASTRAK



Stand on a flat surface

Figure 2 A radon detector

# Part B Investigating radon

# How to detect radon

You can make a radon detector with a yoghurt pot, Blu-Tack, clingfilm and TASTRAK. It needs to be left in position for at least a week to record a reasonable number of tracks.

1 Get your TASTRAK ready. When you are ready to do the experiment, take the TASTRAK carefully from its 'radon proof' bag. Note that TASTRAK is brittle.

You will need to mark the TASTRAK to identify it as yours. Decide what code you will use.

Quickly but carefully use a compass point or pin to scratch *your* code on the TASTRAK surface at one end.

You can then put your TASTRAK inside a plastic bag while you prepare the site for the experiment. If you take it home, wrap it in a plastic bag and use cooking foil on the outside. (Do NOT wrap it in paper – paper is quite radioactive.)

- 2 Make your radon detector. Remove the TASTRAK from its plastic bag and fix it to the bottom of the yoghurt pot with 'Blu-Tack'. The side scratched with your code must be facing you as you look into the yoghurt pot. Cover completely the mouth of the yoghurt pot with a layer of clingfilm and fix it with an elastic band. Radon gets into the pot by diffusing through the sides and the clingfilm.
- **3** Set up experiments. You may set up experiments to measure the activity of radon:

# a On the ground

Turn the yoghurt pot upside down and place it on the ground. Cover it with a large sheet of plastic to keep the rain off. Note the time and date of the start of the exposure.

# **b** In the ground

Dig a hole about 8 cm deep and put in the yoghurt pot (upside down). Cover it with a large sheet of plastic to keep the rain off. Note the time and date of the start of the exposure.

# c In a room

Place the yoghurt pot on a bookshelf or sideboard well away from open windows or sources of heat such as a radiator. Note the time and date of the start of the exposure.

4 Expose the TASTRAK to radon. Leave it for *at least one week*.

- 5 Afterwards. Once the exposure time has finished, remove the TASTRAK carefully. Rinse with distilled water and dry with tissue. Quickly put the TASTRAK into its **metallised** bag and seal with Sellotape. Note the time and date at the end of the exposure.
- **6 Post for processing.** Pack the TASTRAK sample(s) in a well padded bag and return the sample(s) to the processing address given in the Teachers' Notes.

# Analysing your results

When the TASTRAK has been returned from processing

### **1** Analyse the TASTRAK

**a** Put the TASTRAK and scale on the microscope stage. Adjust the magnification and focus so that you can see the tracks clearly. The tracks are up to 30 micrometres in size.

**b** Use the scale to work out the area of your field of view – that is the area you can see. (Ask for help if you have difficulty.)

**c** Count the number of tracks you can see. Move the TASTRAK so that you are looking at another part of it. Count the number of tracks again. (It is best to choose areas to count at random – say a minimum of 10 and find an average.) Work out the average number of tracks per square centimetre of your TASTRAK.

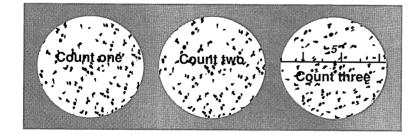


Figure 3 Counting the tracks through the microscope

2 Calculate the radon activity using this equation.

Radon activity (in Bq/m<sup>3</sup> of air) =

 $\frac{5.7 \times (\text{tracks per sq cm})}{\text{exposure time in days}}$ 

- 3 Collect together all the results.
- 4 Write a report. In your report describe why you planned your investigation as you did as well as presenting your results. Are there any readings which you think should be repeated or any localities which should be investigated in greater detail? If so explain. What advice can you give from your results?

You will need • a microscope with

- magnification × 200
- a scale to calibrate the field of view



Figure 4 Working out the average

If you would like to add your results to a national survey of radon levels:

 find the six figure National Grid reference of the place where you took your reading from an Ordnance Survey map. Note it down.

A form for entering your individual or class results is provided. See that your result is entered correctly.

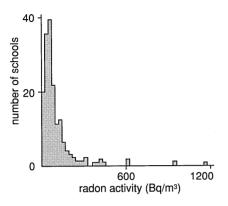


Figure 5 The data for radon activity on the surface of the soil (see table) drawn as a histogram

# Part C – Looking at data

Since 1987, schools have carried out experiments similar to those described in this unit. The table gives their results for radon in the soil.

Type of experiment	Radon activity in Bq/m³	Number of samples
On the surface	$114 \pm 30$	159
8 cm below the surface	$460 \pm 90$	333

These are averages. In fact, the range of results they got was very much larger, often from  $30 \text{ Bq/m}^3$  to  $1000 \text{ Bq/m}^3$ . Some values were higher than this.

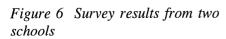
Two schools, Portway School (Bristol) and Brookside Junior School, Street (Somerset) carried out surveys in homes.

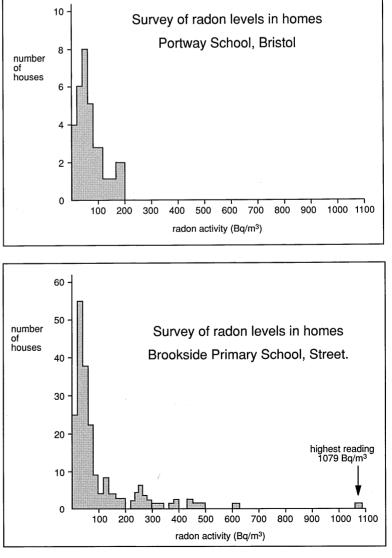
Q1 Look at the table. Suggest why the radon activity on the surface of the soil was lower than that at 8 cm below the surface.

**Q2** Describe the pattern shown by the histogram in your own words.

Did you collect similar data in your own investigation? If so, comment on where your results fit in.

Q3 Look at the results of the two schools' surveys shown in Figures 5 and 6. Comment on each set of results. Would you recommend any householders to 'radon-proof' their homes?





# page 4

# **Further information**

# **Radon daughters**

Although this unit focuses on 'radon', 99 per cent of the radiation you receive through radon is due to the 'decay products' of radon – those are other atoms further down the uranium decay series. They are also alpha emitters but with very short half-lives and are known as 'radon daughters'.

#### Extra questions

**Q4** Look up the meaning of the following terms which have been used in the unit and make yourself a glossary. (A glossary is an alphabetical list of words or phrases with their definitions or explanations.)

alpha particle a becquerel decay chain half-life isotope radioactive radioisotope

**Q5** Radon can diffuse through the clingfilm and pot. Explain why the yoghurt pot detector detects alpha particles from radon and radon daughters only (and not those of other radioisotopes).

#### The National Radiological Protection Board (NRPB)

The NRPB is an independent statutory body. It provides advice and services to government, local authorities, industry, environmental groups, trade unions etc.

If you would like to know more about radon in the environment you can get free leaflets such as *Radiation Doses – Maps and Magnitudes* and *Radon – Questions and Answers* from

The National Radiological Protection Board The Information Officer Chilton Oxfordshire OX11 0RQ

*The Householders' Guide to Radon* is published by the Department of the Environment – available from HMSO and booksellers.

*Living with Radiation* is published by NRPB – available from HMSO and booksellers.

#### Facts about alpha radiation

 It consists of alpha particles (two protons and two neutrons).

 Alpha particles carry a charge of +2e.

 $\odot$  Their speed is around 10' m/s in air.

 $\odot$  They travel no more than 5 cm in air.

 They cannot pass through paper, metal or the skin covering the outside of your body.

 when they hit living tissue they produce ions which can change or destroy living cells. Changed cells may cause cancer.

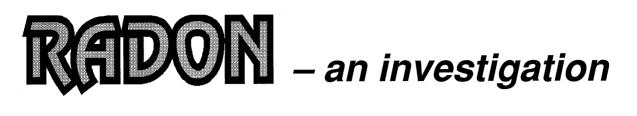
# **Properties of TASTRAK**

 TASTRAK records all naturally occurring alpha particles, low energy protons and heavy ions.

 TASTRAK is not affected by light, gamma-rays, X-rays, betaparticles and sea-level cosmic rays.

 TASTRAK is easy and safe to handle whether exposed to radiation or not.

Answers to the questions are in the *Teachers' Notes*.



# The results of our survey are

Grid reference (6 figure number)	Radon level (Bq/m³)	Location (soil/room)	Dates	Name of student(s)
-				
			-	

	Teacher's Name			
Post to	School			
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